

Calcium Metabolism

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Learning Objectives

- Forms of calcium and factors affecting its analysis
- Regulation of calcium with its molecular mechanisms and applications
- Functions of calcium, mechanism of action and clinical application
- Clinical Significance

CALCIUM

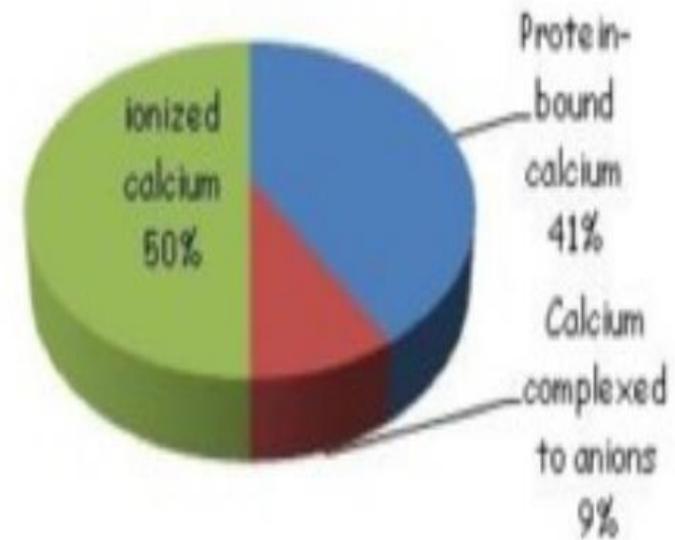
Total Body Content : 1-1.5kg

Bones : 99%

ECF : 1%

Plasma : 9-11mg/dl

Distribution of Body calcium



Tissue distribution and states of calcium

Tissue	Calcium
Skeleton	99%
Soft tissues	1%
Extracellular fluid	<0.2%
Total	1000 g (25 mol)

Free (ionized)	50
Protein-bound	40
Complexed	10
Total (mg/dL)	8.6-10.3
(mmol/L)	2.15-2.57

Daily Requirement and Dietary sources

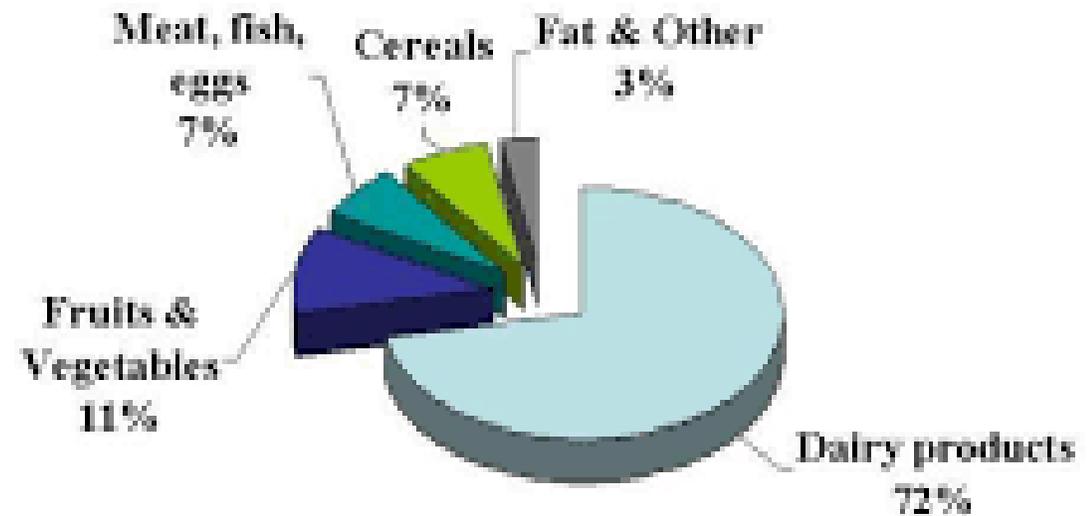
Adults : 800mg

Children : 1200mg

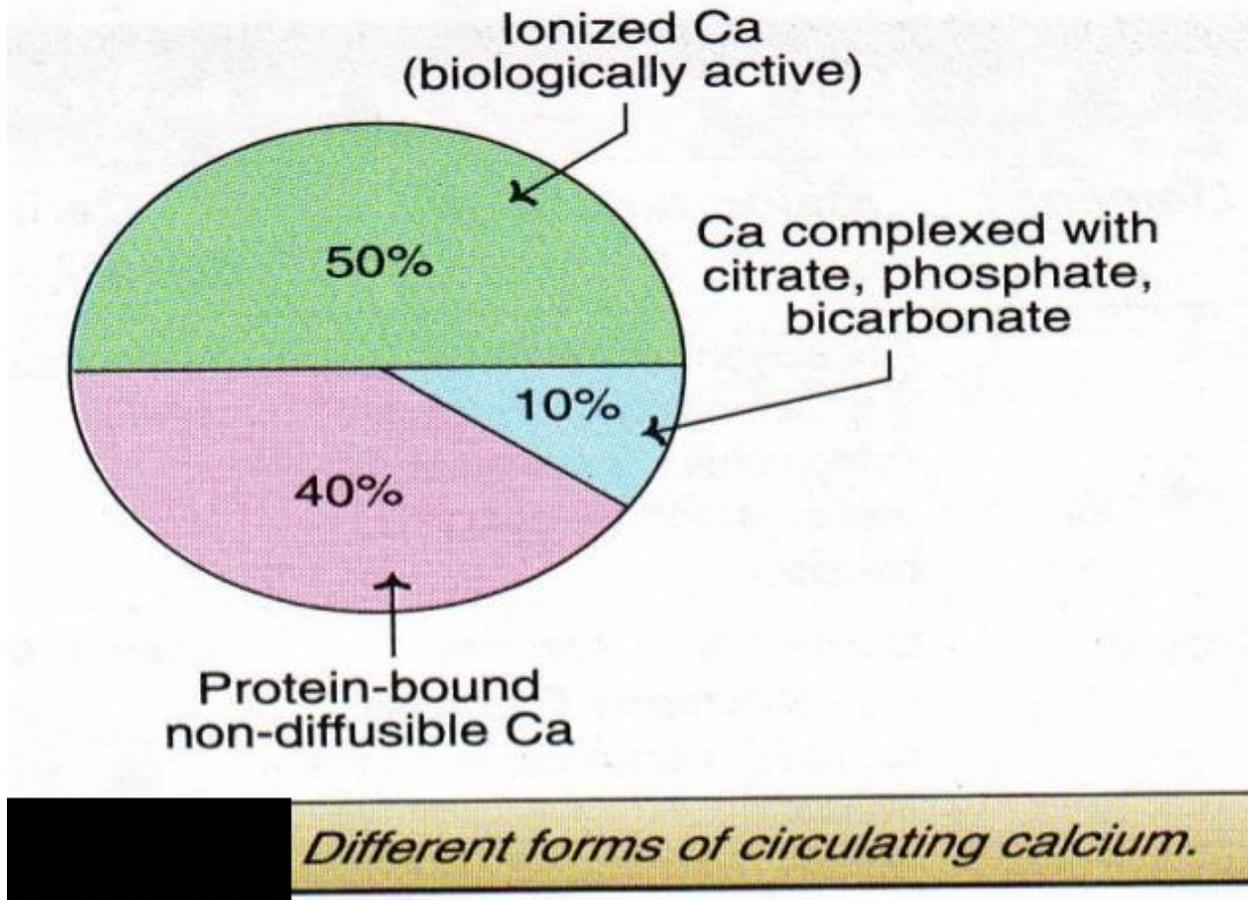
Preg. & Lactation : 1500mg

Postmenopausal- 1500 mg

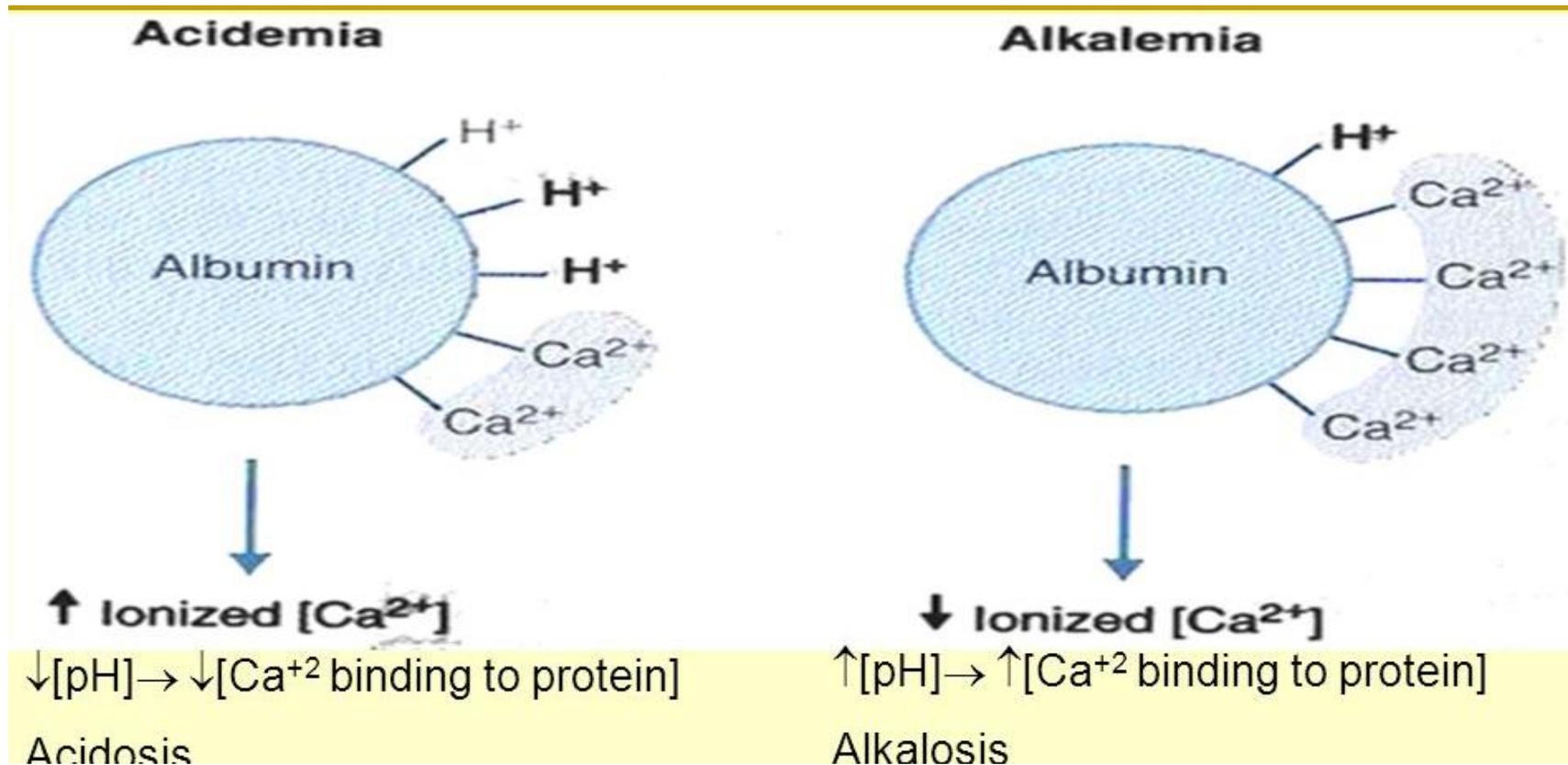
Dietary sources of Calcium



Forms of calcium



Effect of albumin concentration and acid base balance on plasma concentration of ionised calcium



Adjusted Calcium for Hypoalbuminemia

Total calcium (mg/dL) corrected for hypoalbuminemia =
Total calcium (measured) + [(Normal albumin –
Patient's albumin) × 0.8]

***Corrected Total Calcium (mg / dL) =
Total Calcium (mg / dL) + 0.8 (4 - Albumin [g / dL])***

Free calcium: ISE
More useful

Pre analytical factors for measurement of calcium

Torniquet application: increased total not ionized Calcium

Fist Clinching/forearm exercises: increased free calcium.

Posture: standing increase total calcium

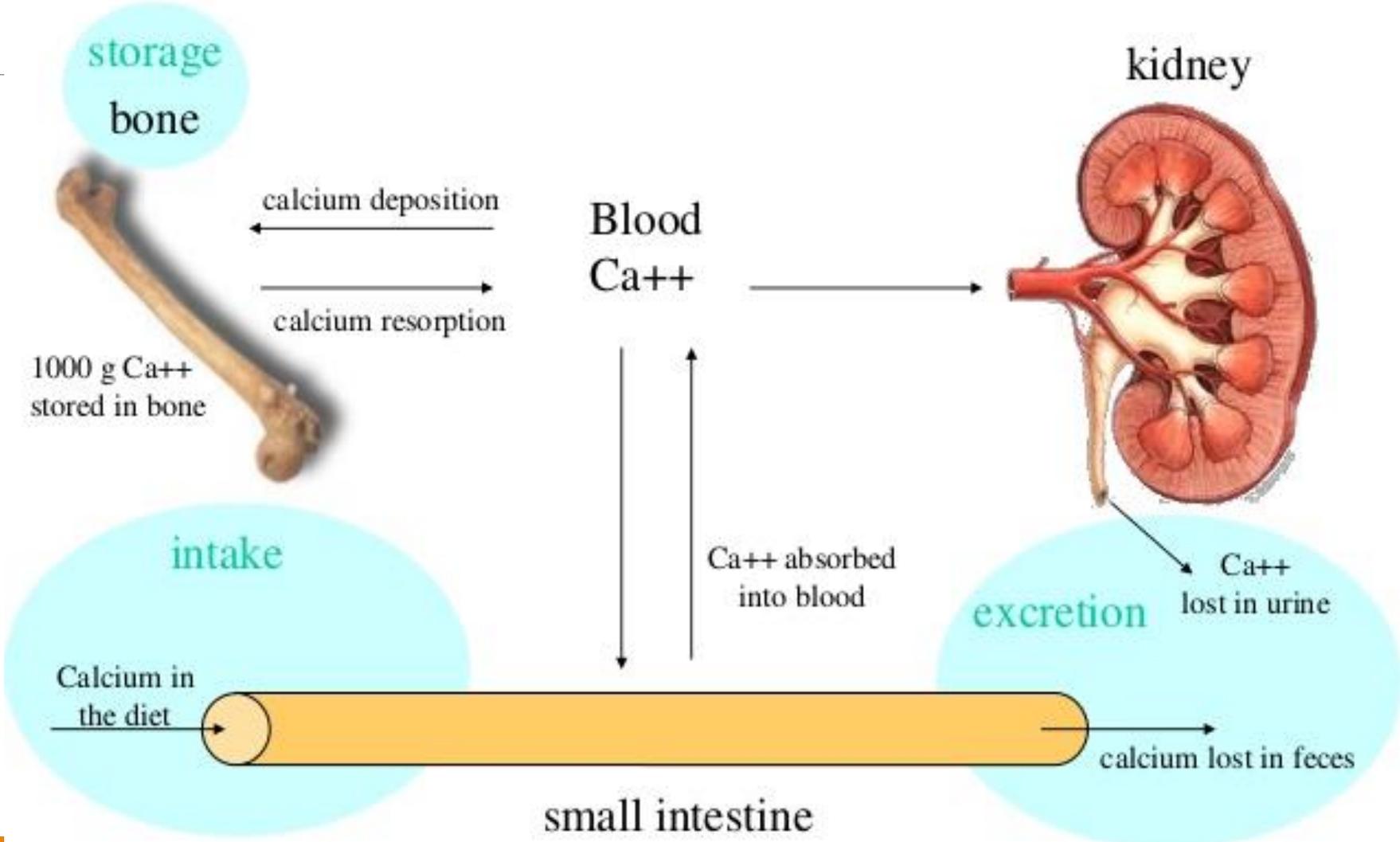
Prolonged immobilization and rest: decreased bone density and increased free calcium

Physiological Variation:

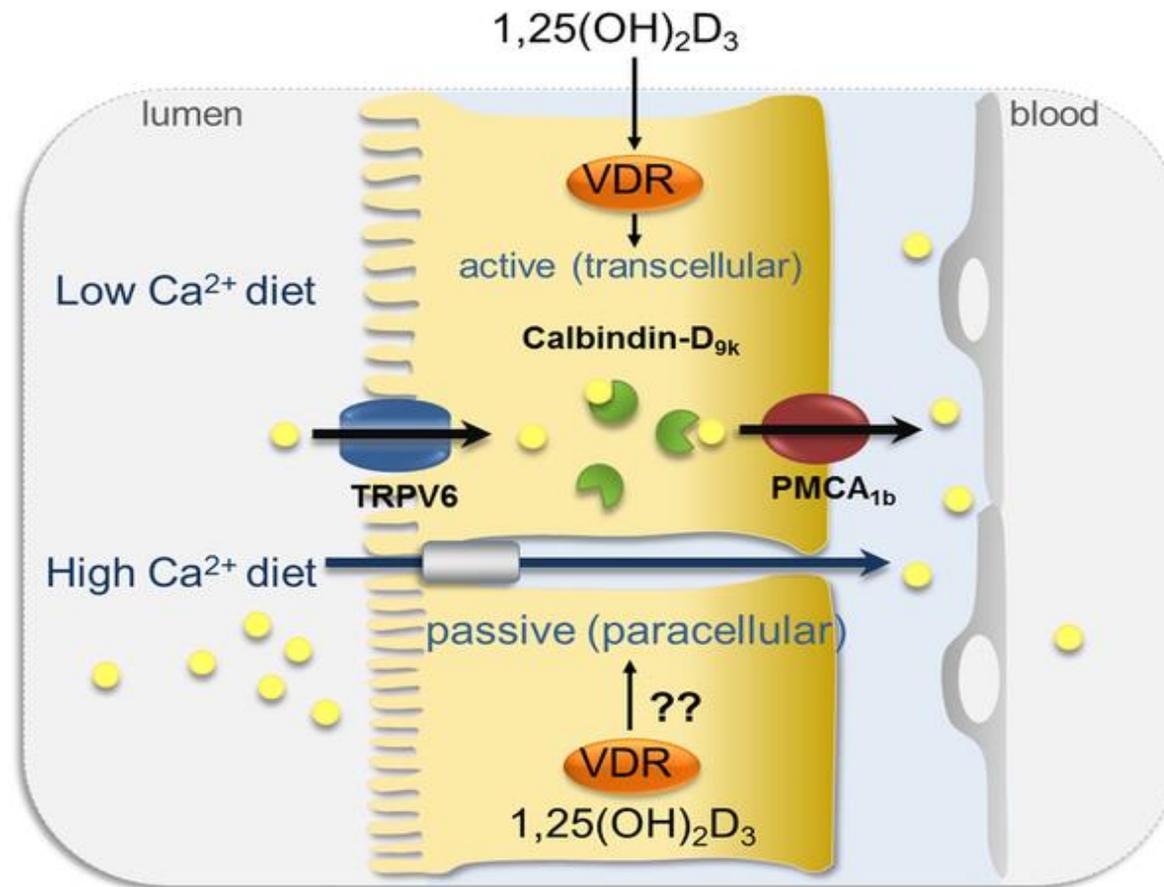
Age

Pregnancy: decreased total calcium

Calcium homeostasis



Calcium absorption in intestine



Factors Promoting Absorption

Vit.D: calcitriol induces synthesis of carrier protein (calbindin) in intestinal epithelial cells

PTH: increases synthesis of calcitriol

Acidic pH of intestinal milieu makes Ca salts soluble

Factors Decreasing Ca Absorption

Alkaline pH

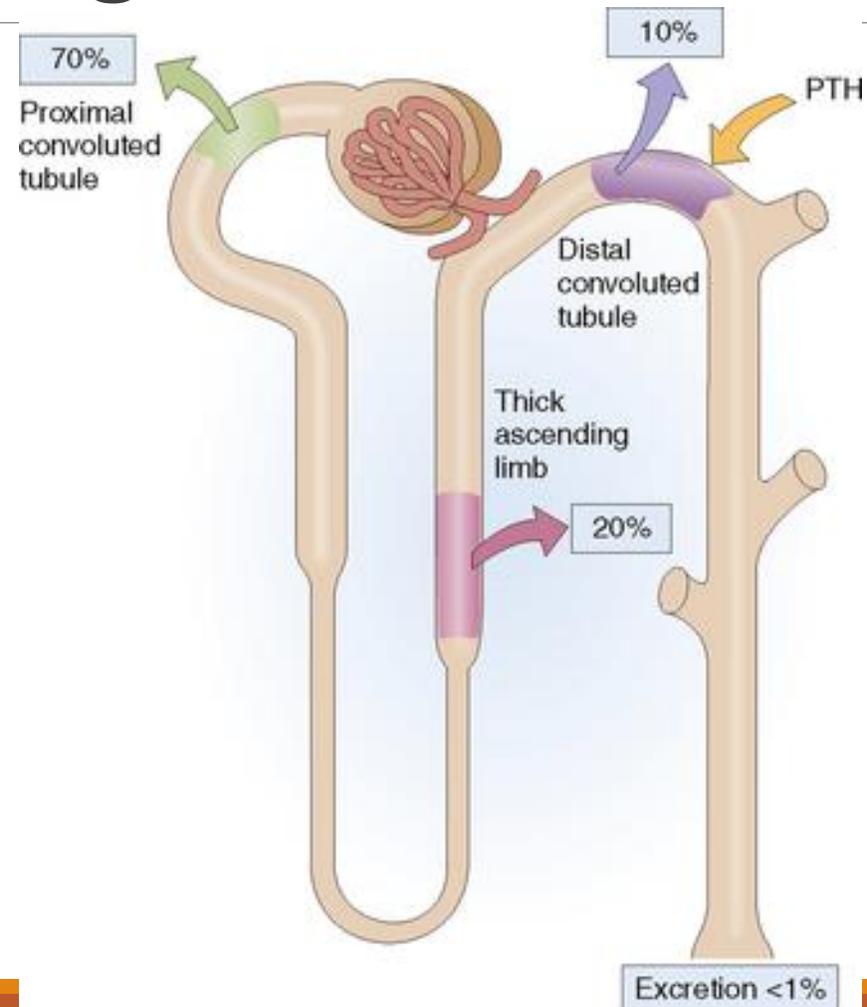
Phytates & oxalates

Malabsorption syndrome: fatty acids are not absorbed forming insoluble Ca soaps

Excess of dietary fibres

Excess of phosphates (1:2 to 2:1)

Renal handling of calcium



Control of Calcium Excretion By the Kidneys

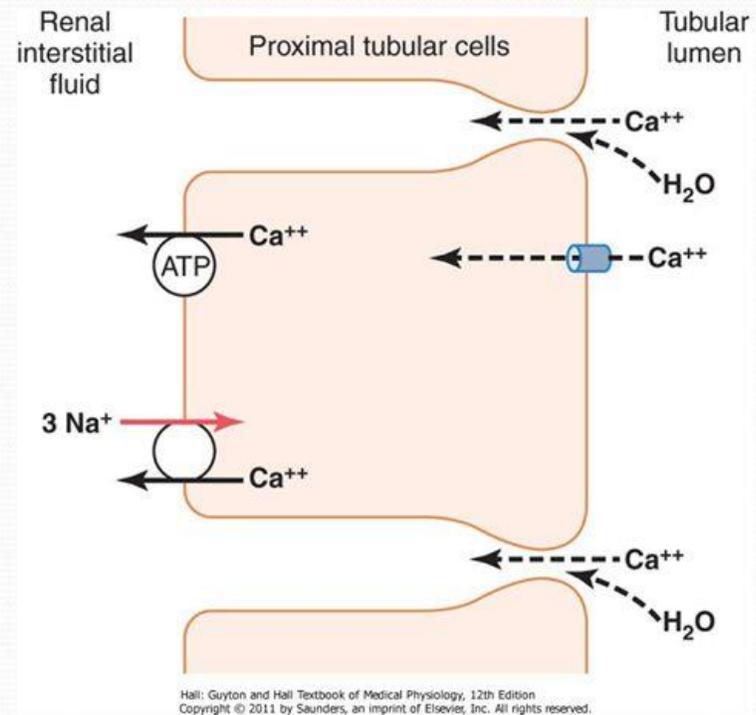
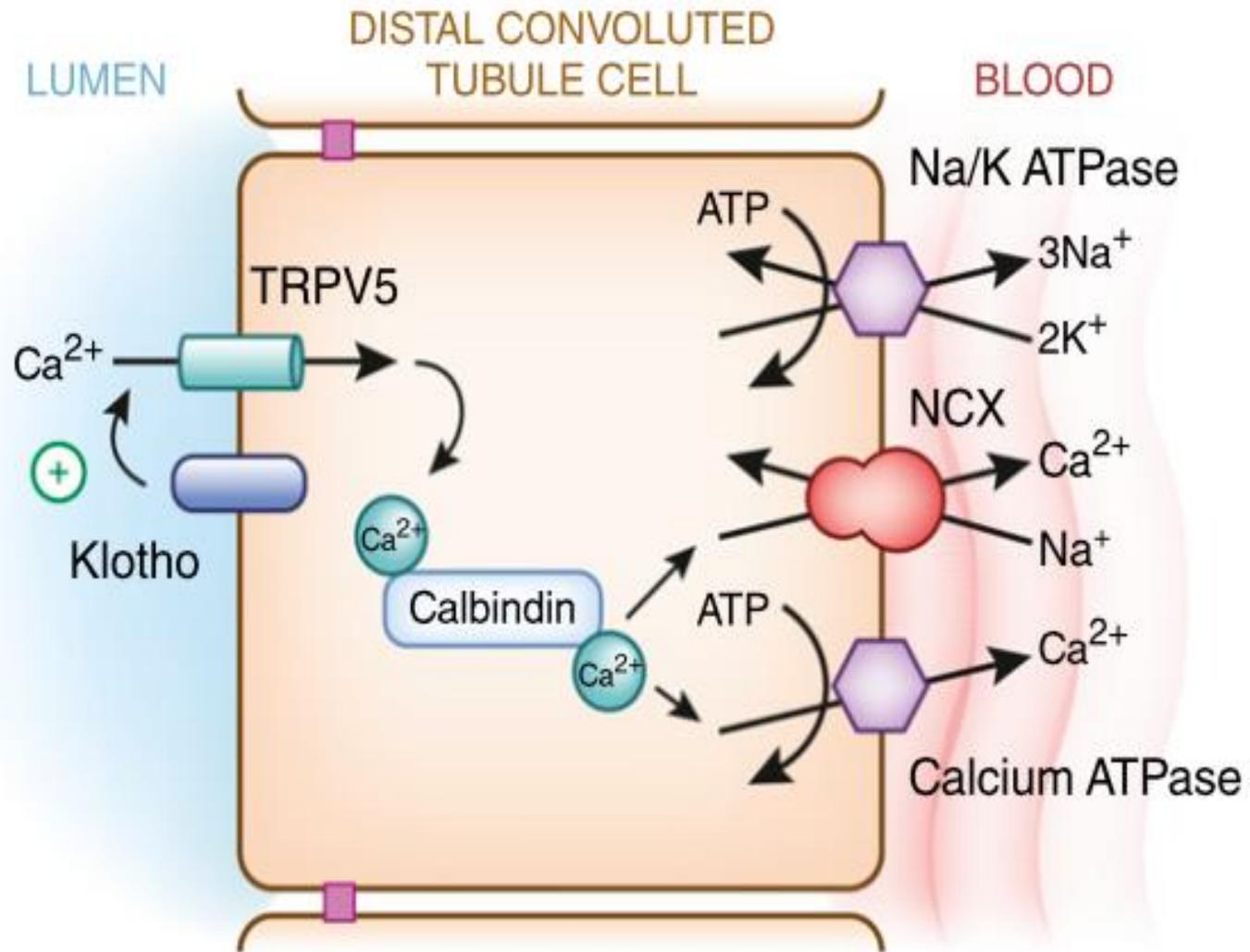
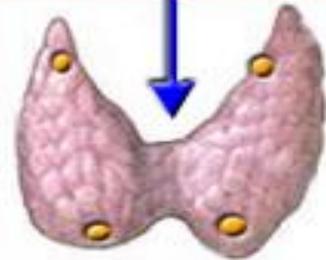


Fig. 29.12 Mechanisms of calcium reabsorption by paracellular and transcellular pathways in the proximal tubular cells



LOW BLOOD CALCIUM DETECTED AT PARATHYROIDS

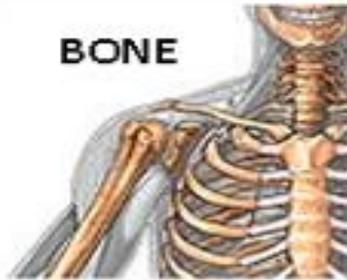


PARATHYROIDS

PTH



BONE



PTH stimulates osteoclast activity. Bone is resorbed which releases calcium into the bloodstream

KIDNEY



PTH stimulates:
1) reabsorption of calcium from urine (and excretion of phosphates)
2) activation of Vitamin D

Active Vitamin D

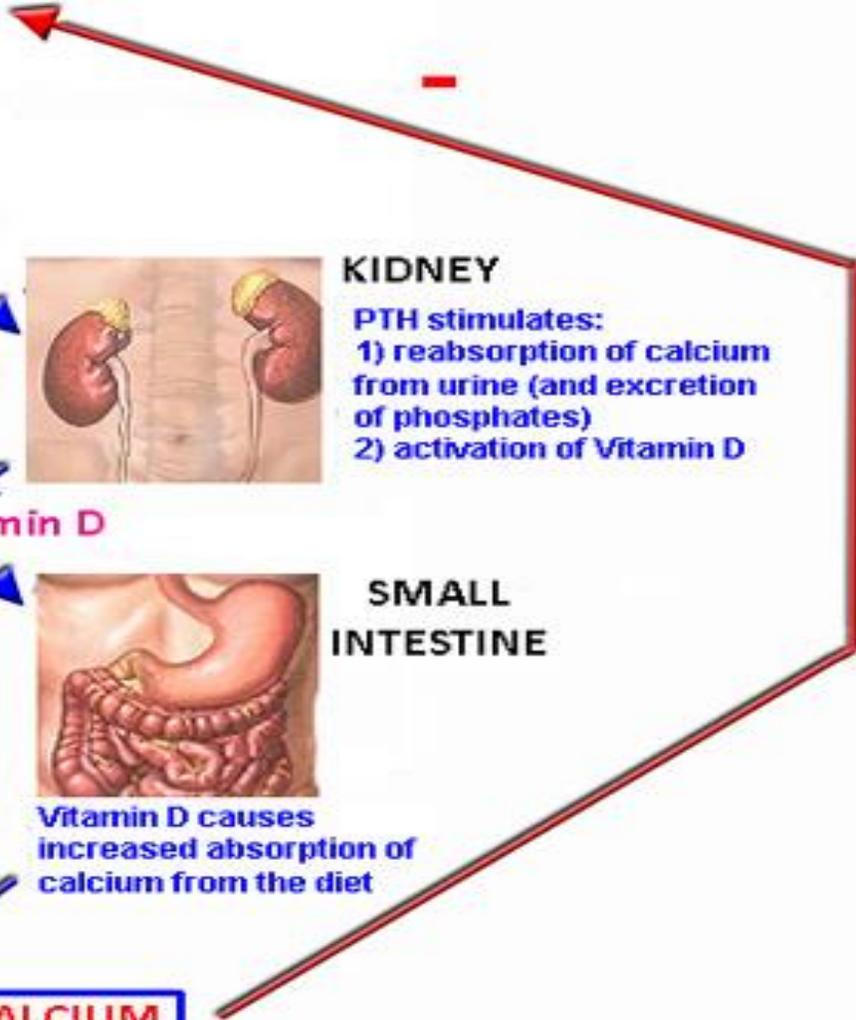


SMALL INTESTINE

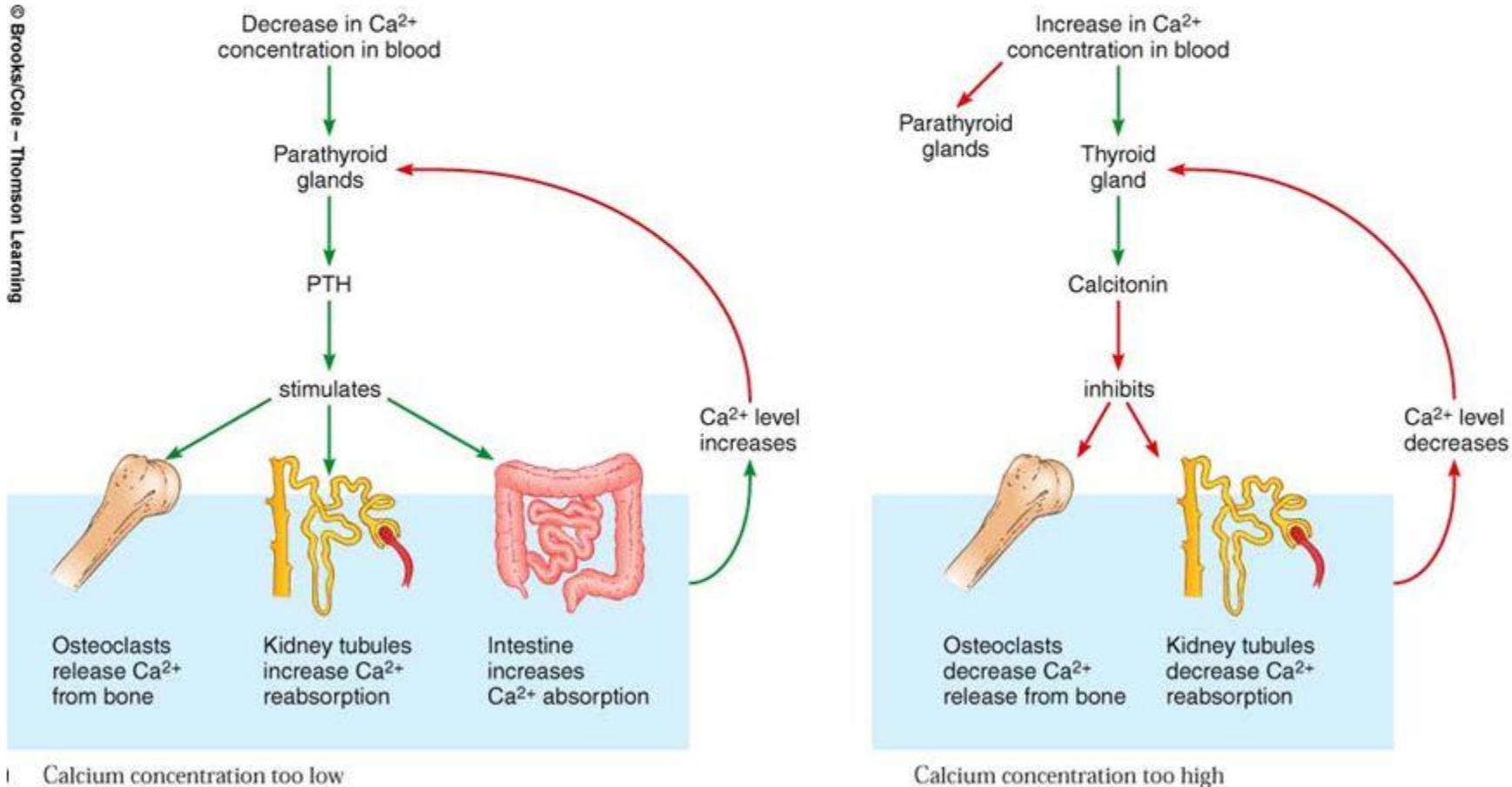


Vitamin D causes increased absorption of calcium from the diet

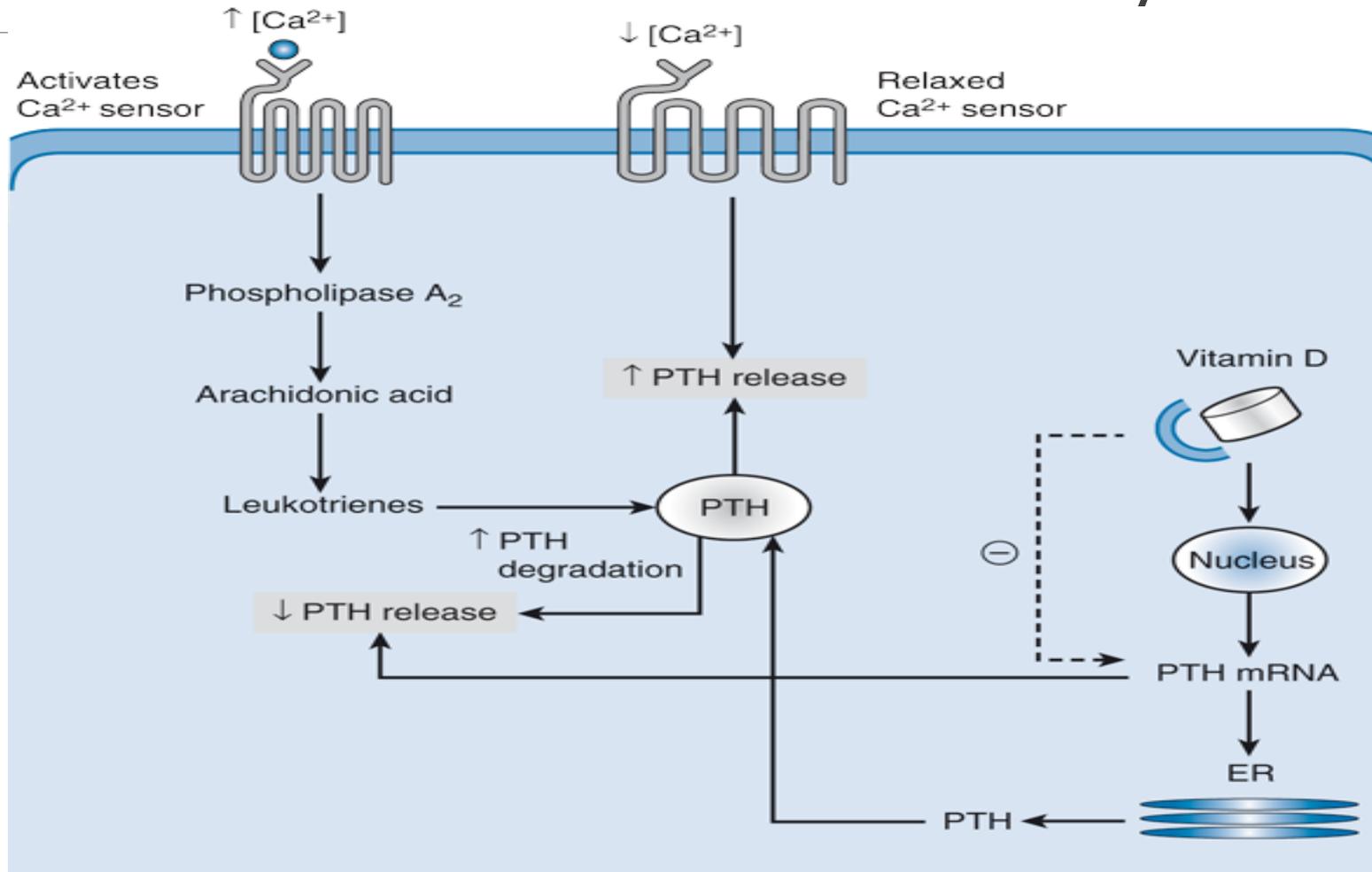
INCREASED SERUM CALCIUM



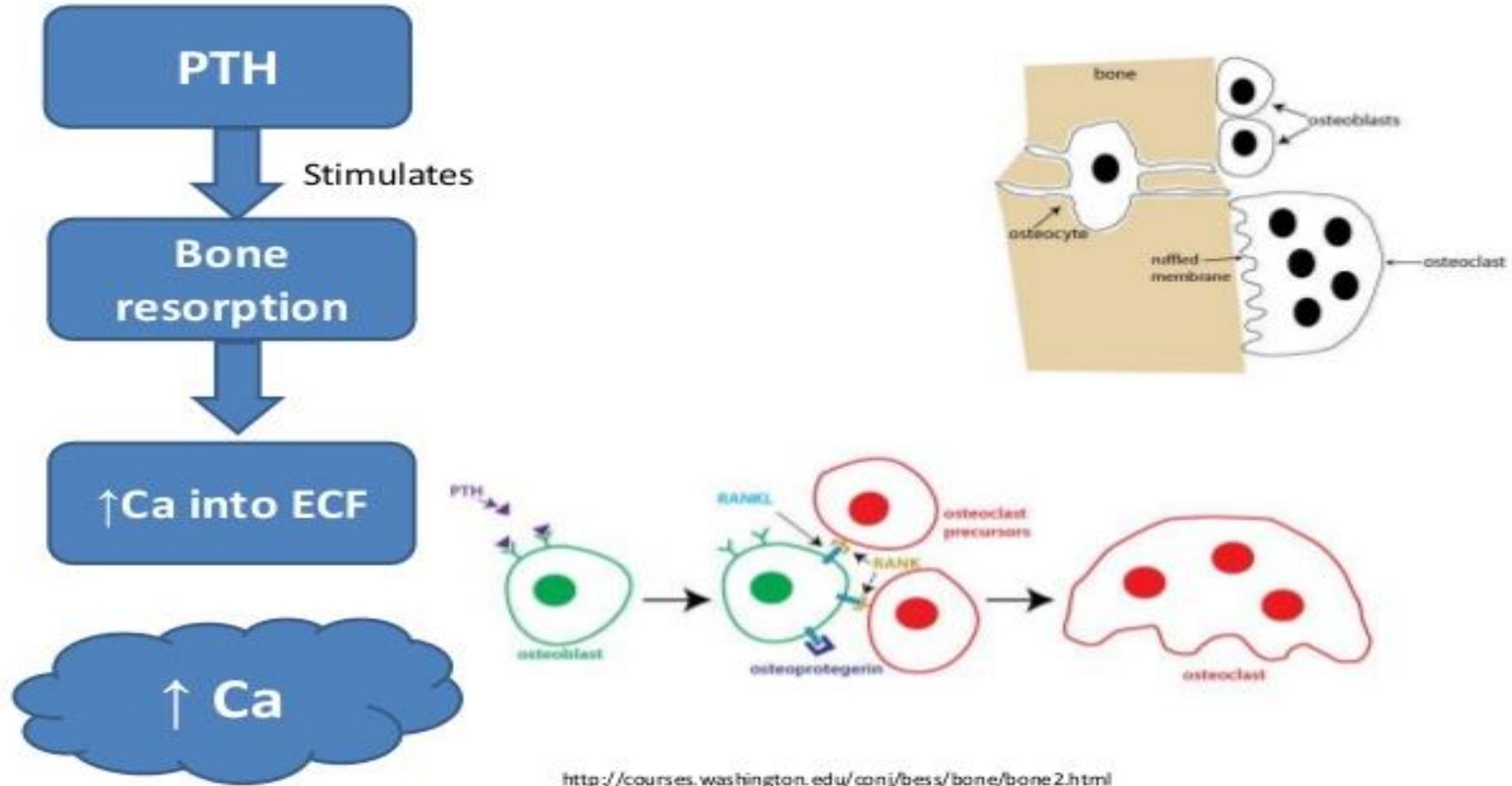
Regulation of calcium homeostasis by PTH and calcitonin



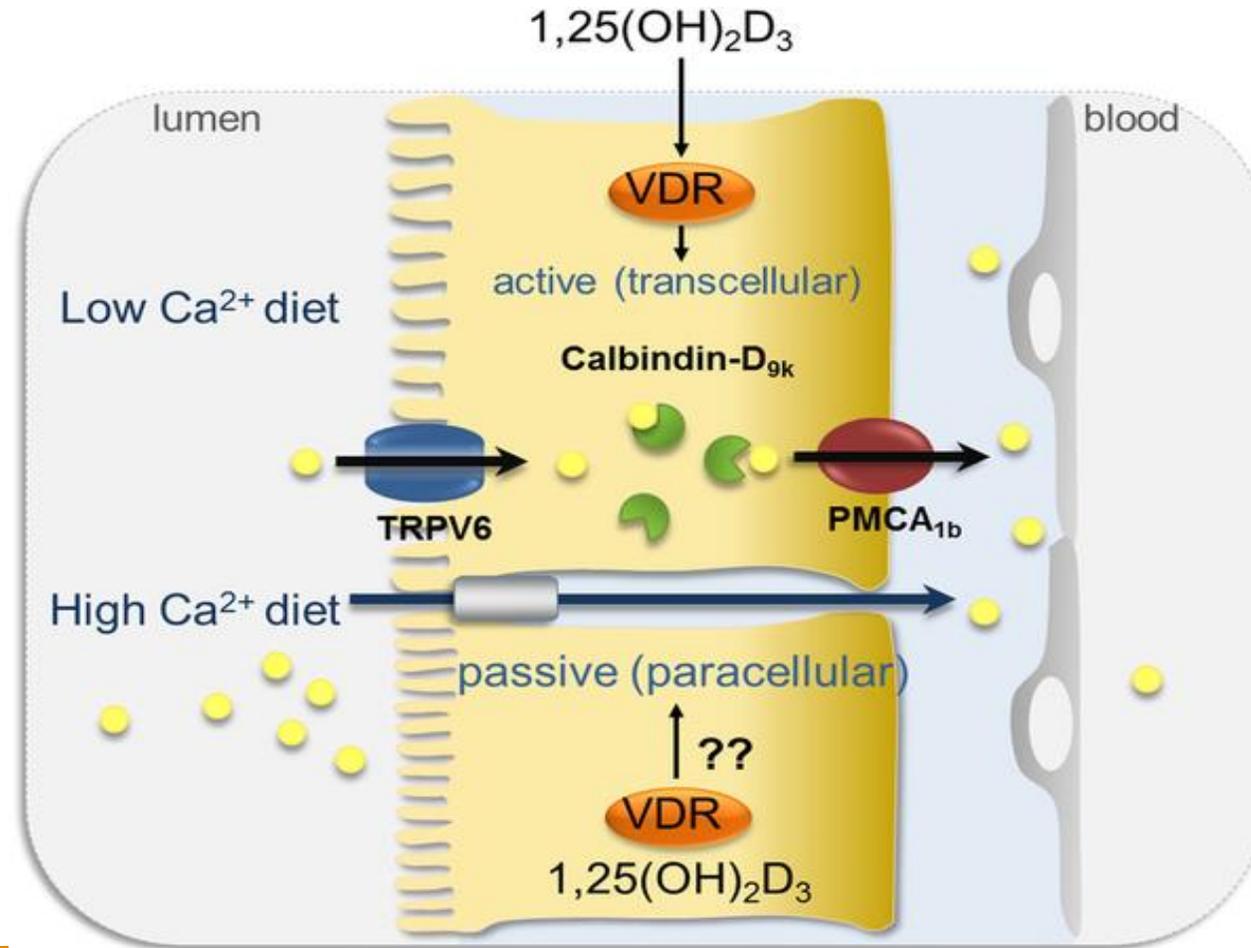
Regulation of PTH secretion by Ca



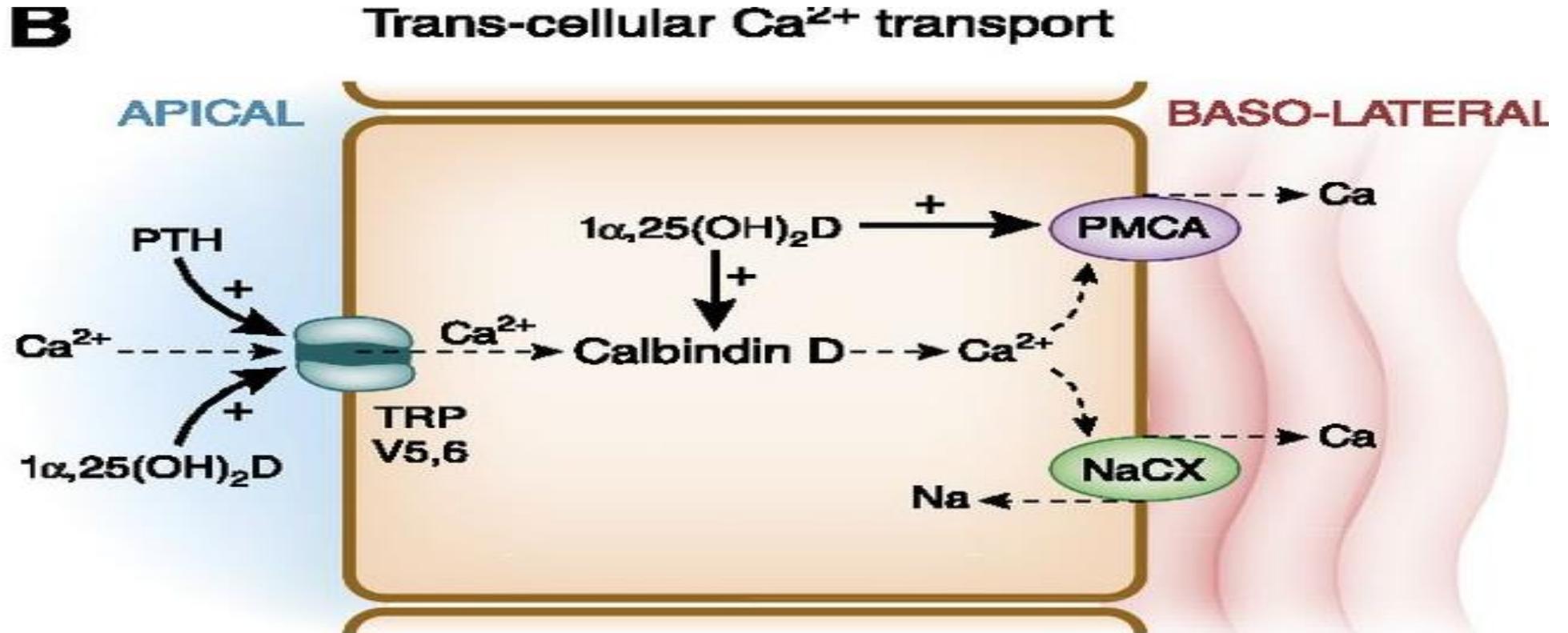
PTH Effects on Bone



Calcium absorption



Calcium reabsorption from kidneys



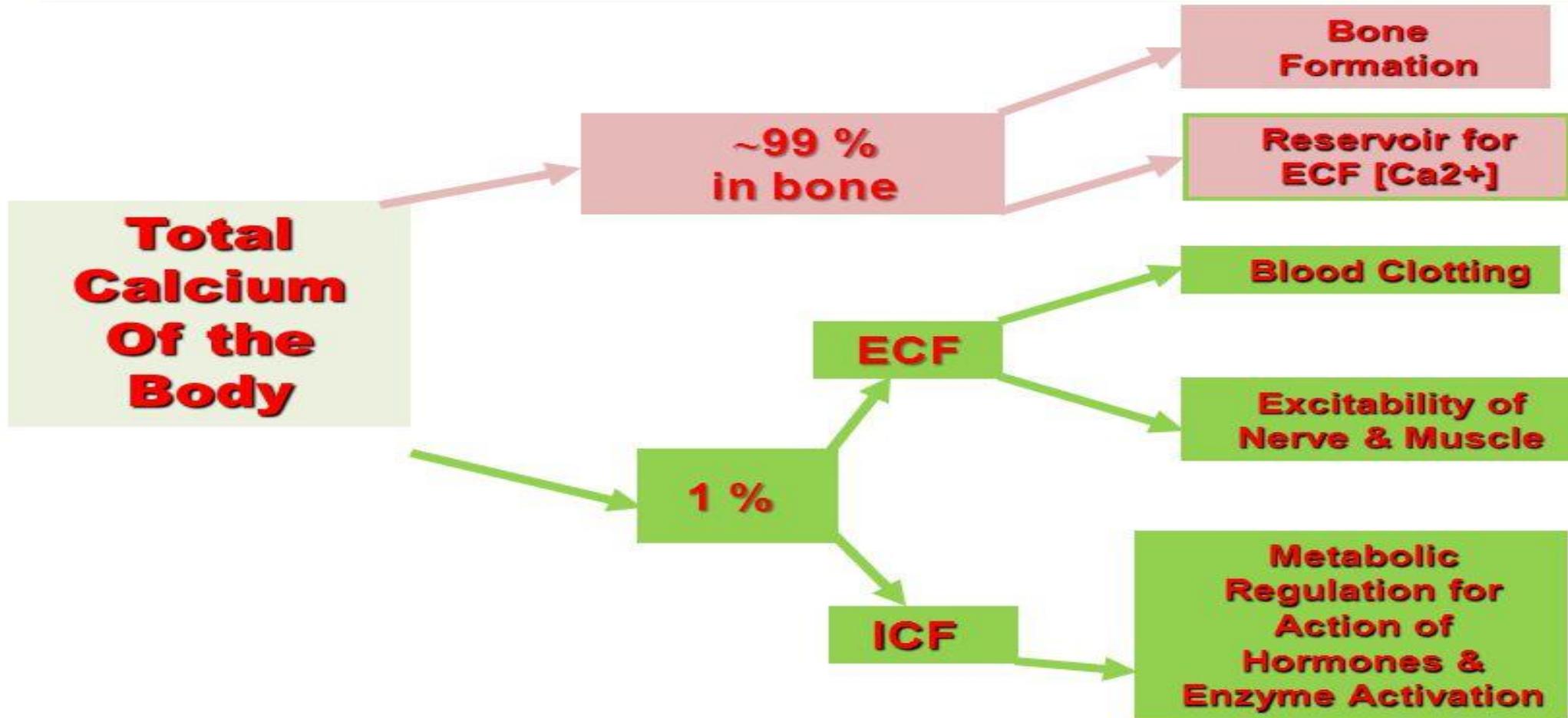
Hypercalcemia of malignancy?

PTHrP?

RANKL?

Denosumab is being clinically used for the treatment of osteoporosis and cancer-related bone disorders

Biological Functions of Calcium



Preserves bone density

Chemical Composition of Bone

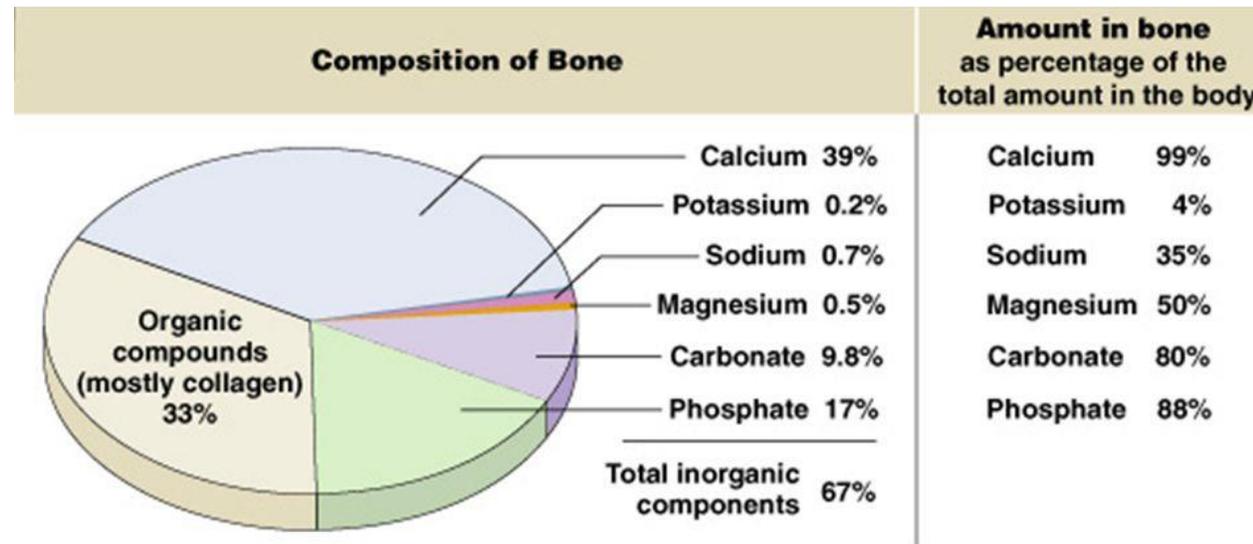
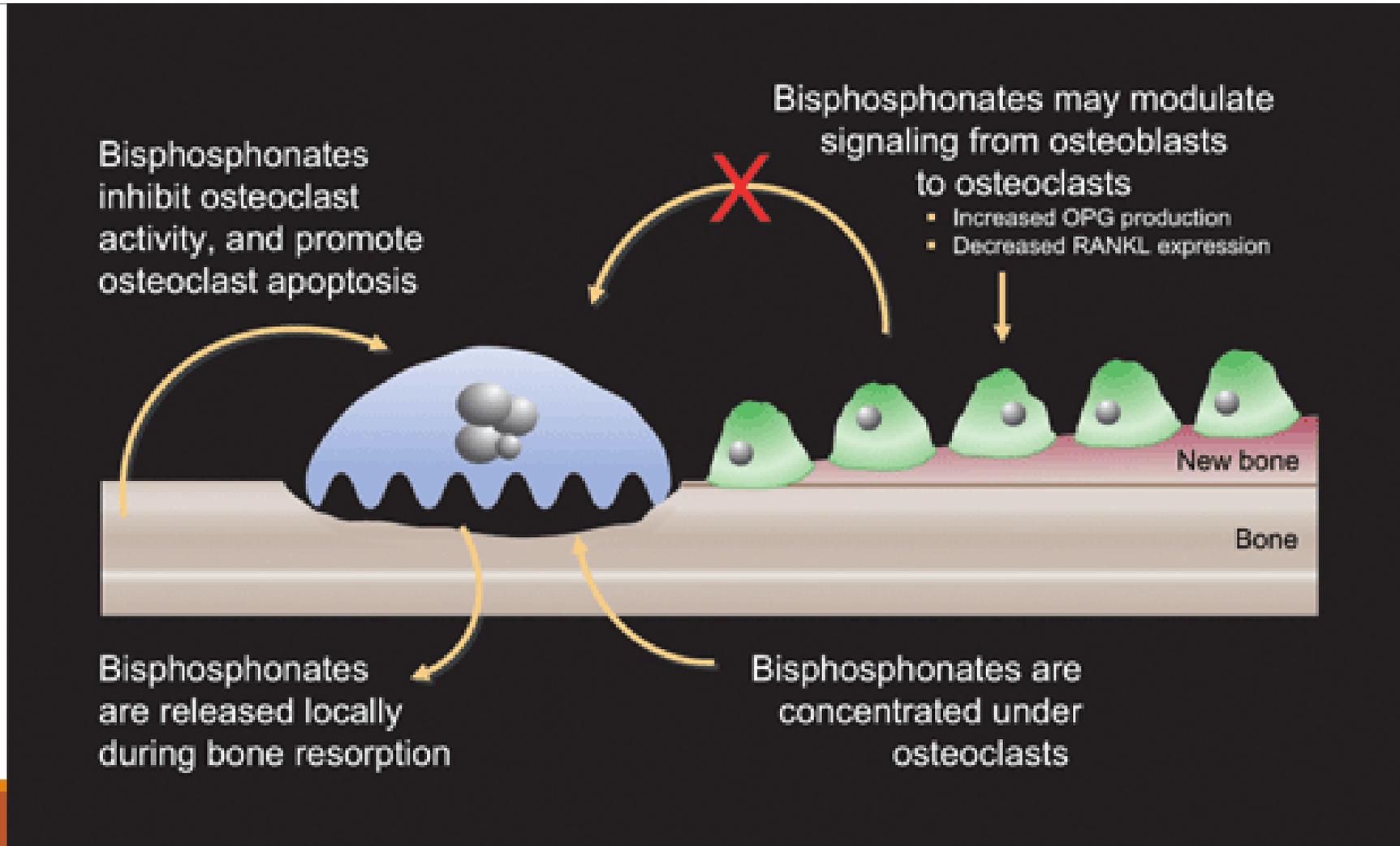
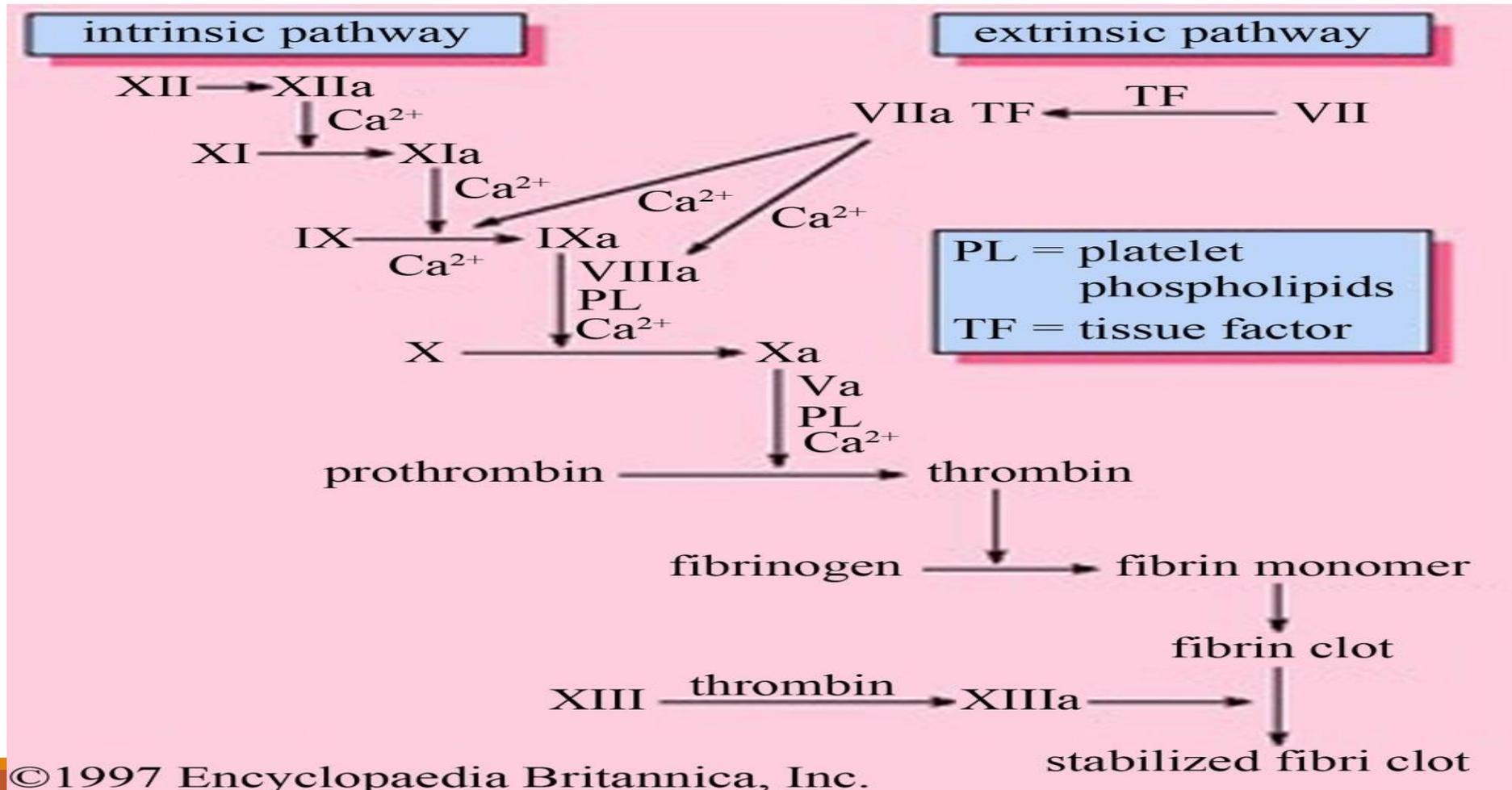


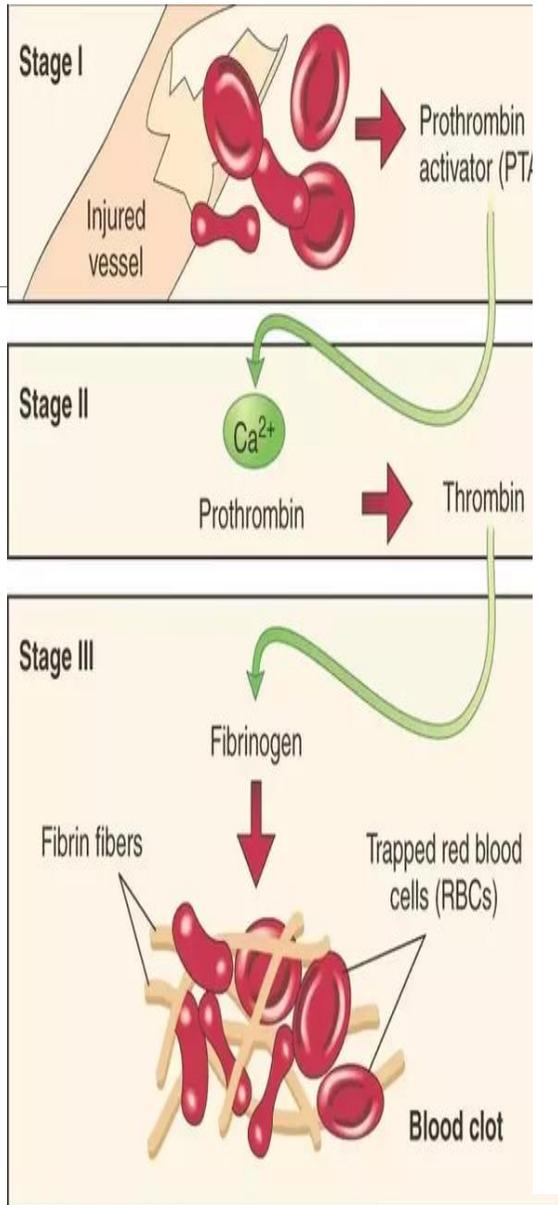
Figure 6-13

Mechanism of action of Bisphosphonates



Role of Ca in blood clotting





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Role of Calcium in Blood Coagulation

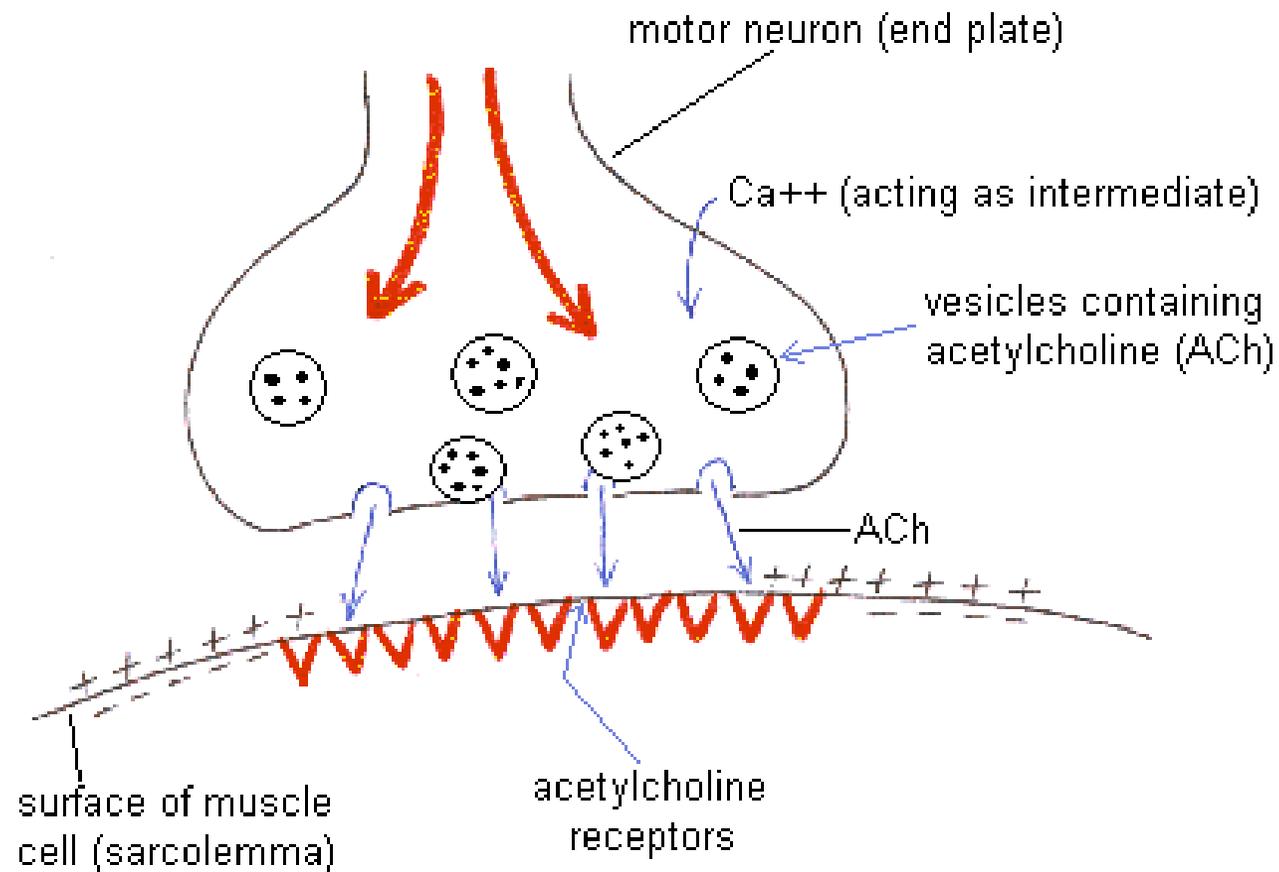
Clotting factors (thrombin, VII, IX and X) contain a unique modified glutamate residue, called carboxyglutamate (Gla).

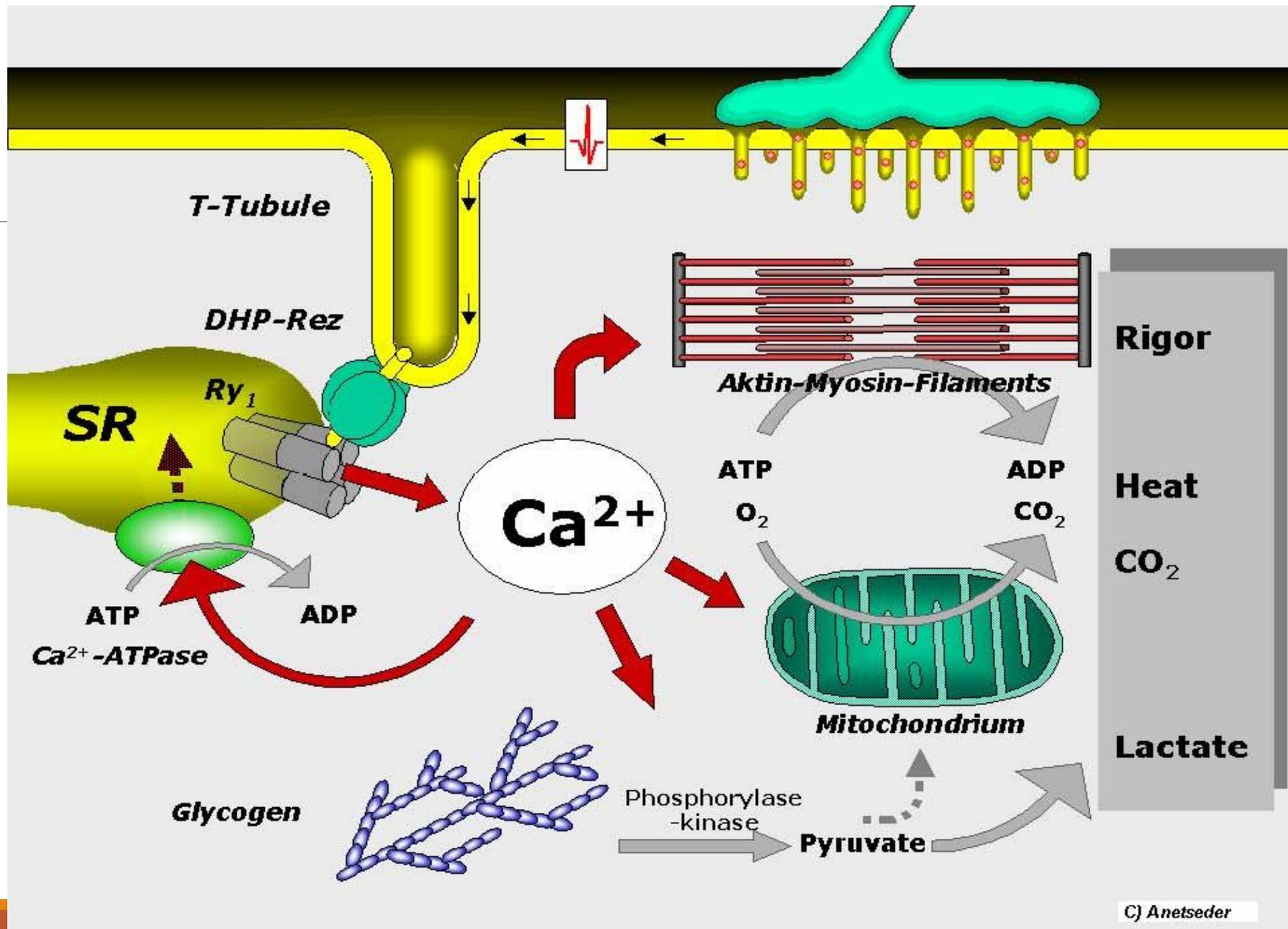
Synthesis of these Gla residues results from post-translational modifications of the newly synthesized factors in the liver endoplasmic reticulum by a vitamin K

This amino acid is a natural high affinity binder (or chelator) of calcium ions, hence the designation of calcium as a co-factor in the blood clotting cascade.

Calcium - Gla-factors complex allow specific interactions with acidic membrane lipids that ultimately lead to correct tertiary and quaternary protein structures recognized by other proteins in the pathway.

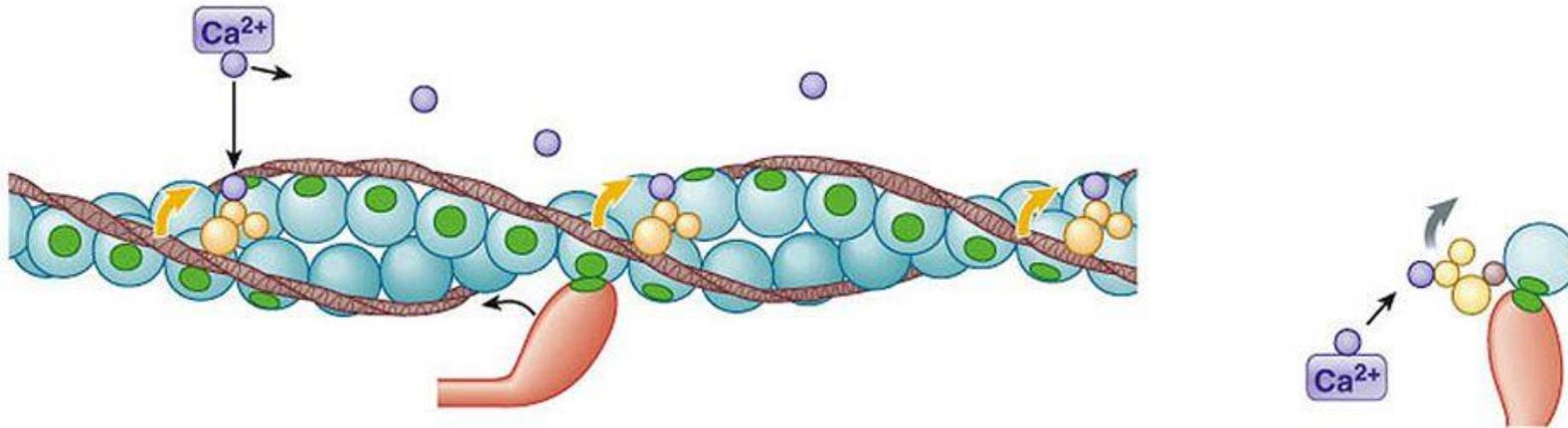
Neuromuscular transmission





Role of Calcium in Cross-Bridge Formation

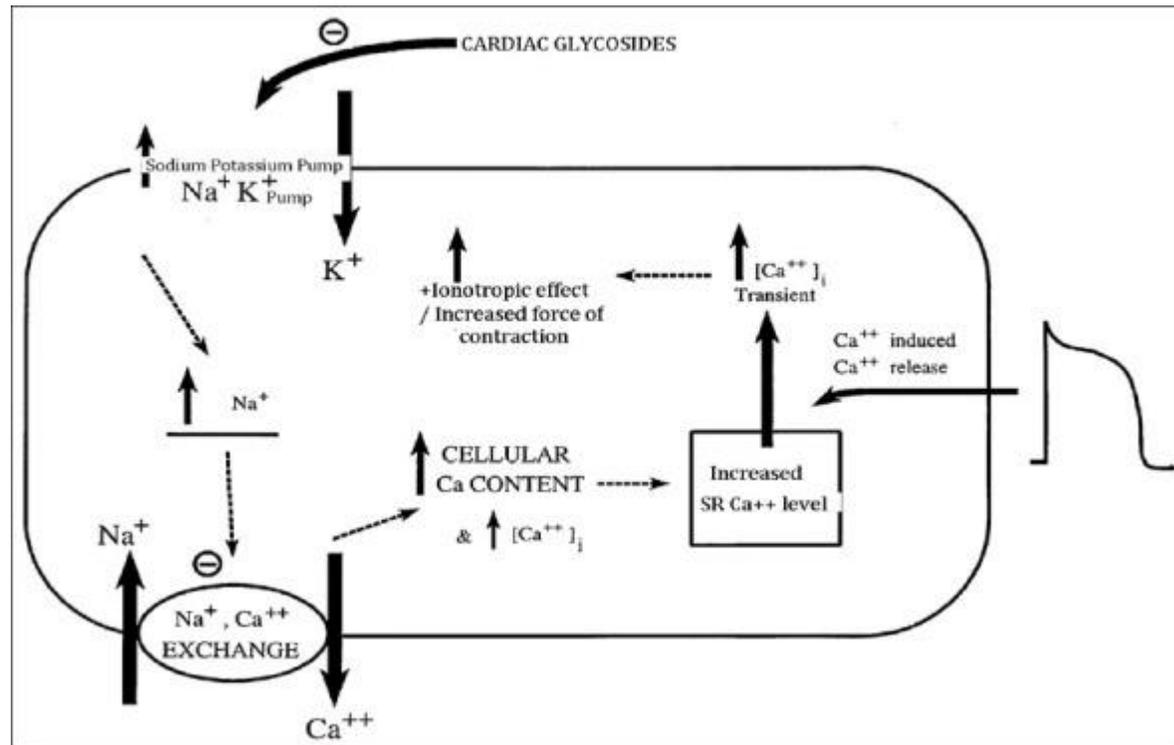
- Excited



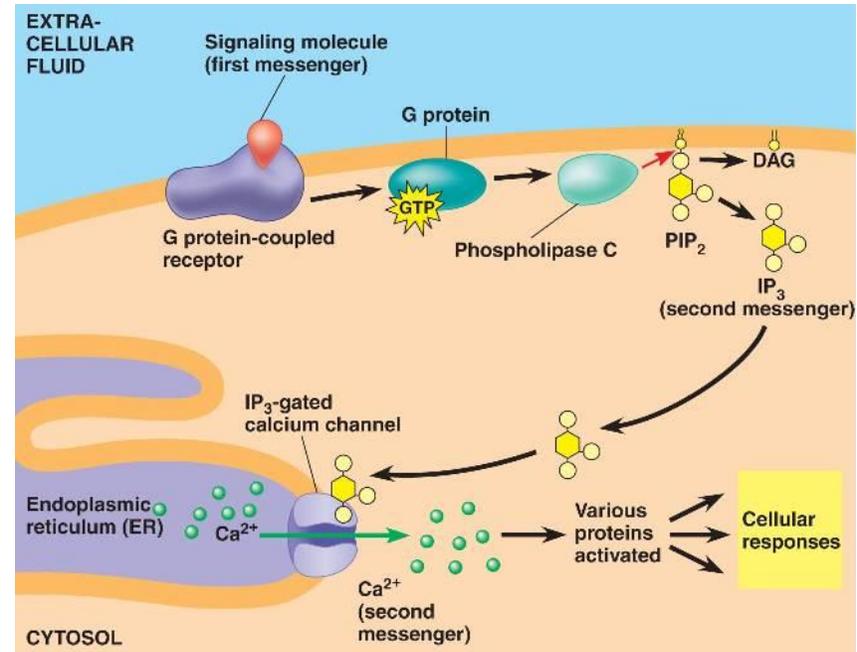
(b) Excited

- 1 Muscle fiber is excited and Ca²⁺ is released.
- 2 Released Ca²⁺ binds with troponin, pulling troponin-tropomyosin complex aside to expose cross-bridge binding site.
- 3 Cross-bridge binding occurs.
- 4 Binding of actin and myosin cross bridge triggers power stroke that pulls thin filament inward during contraction.

Mechanism of action of cardiac glycosides



Calcium as second messenger



The second messenger is calcium or phosphatidylinositols (or both)

Acetylcholine (muscarinic)

α_1 -Adrenergic catecholamines

Angiotensin II

Antidiuretic hormone (vasopressin)

Cholecystikinin

Gastrin

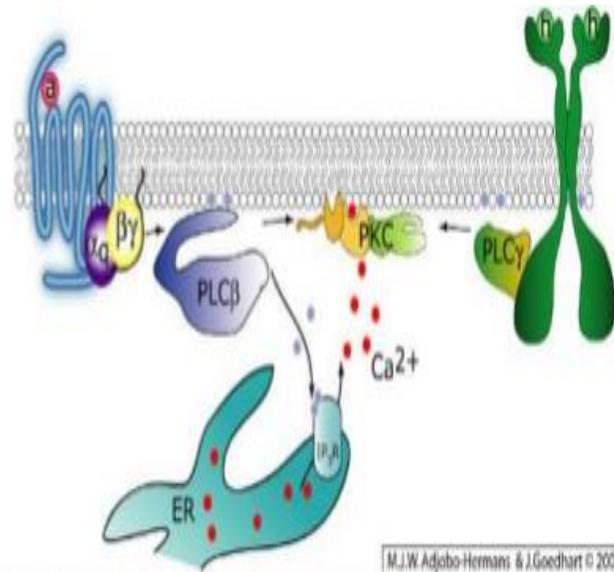
Gonadotropin-releasing hormone

Oxytocin

Platelet-derived growth factor (PDGF)

Substance P

Thyrotropin-releasing hormone (TRH)



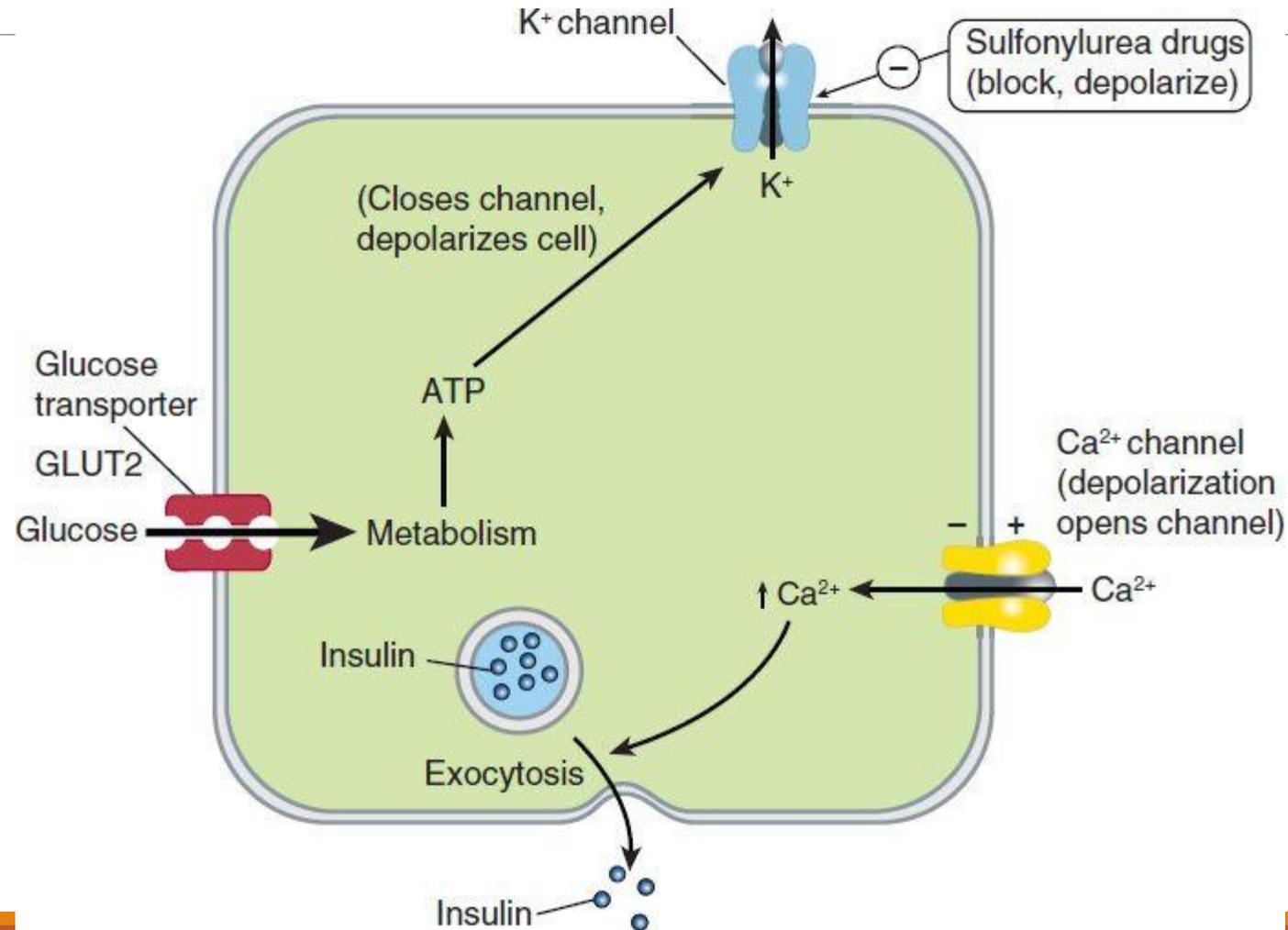
As 2nd messenger

Activation of enzymes- ATPase, Lipase, SDH

**Release of Hormones- PTH,
Calcitonin, Insulin, Vitamin D**

**Secretory process- Endocytosis, exocytosis,
cell motility**

Mechanism of action of Sulfonylurea drugs



Clinical Significance

Hypercalcemia

Hypocalcemia

Hypercalcemia

TABLE 65-1 CAUSES OF HYPERCALCEMIA

Excessive PTH production

- Primary hyperparathyroidism (adenoma, hyperplasia, rarely carcinoma)
- Tertiary hyperparathyroidism (long-term stimulation of PTH secretion in renal insufficiency)
- Ectopic PTH secretion (very rare)
- Inactivating mutations in the CaSR or in G proteins (FHH)
- Alterations in CaSR function (lithium therapy)

Hypercalcemia of malignancy

- Overproduction of PTHrP (many solid tumors)
- Lytic skeletal metastases (breast, myeloma)

Excessive 1,25(OH)₂D production

- Granulomatous diseases (sarcoidosis, tuberculosis, silicosis)
- Lymphomas
- Vitamin D intoxication

Primary increase in bone resorption

- Hyperthyroidism
- Immobilization

Excessive calcium intake

- Milk-alkali syndrome
- Total parenteral nutrition

Other causes

- Endocrine disorders (adrenal insufficiency, pheochromocytoma, VIPoma)
- Medications (thiazides, vitamin A, antiestrogens)

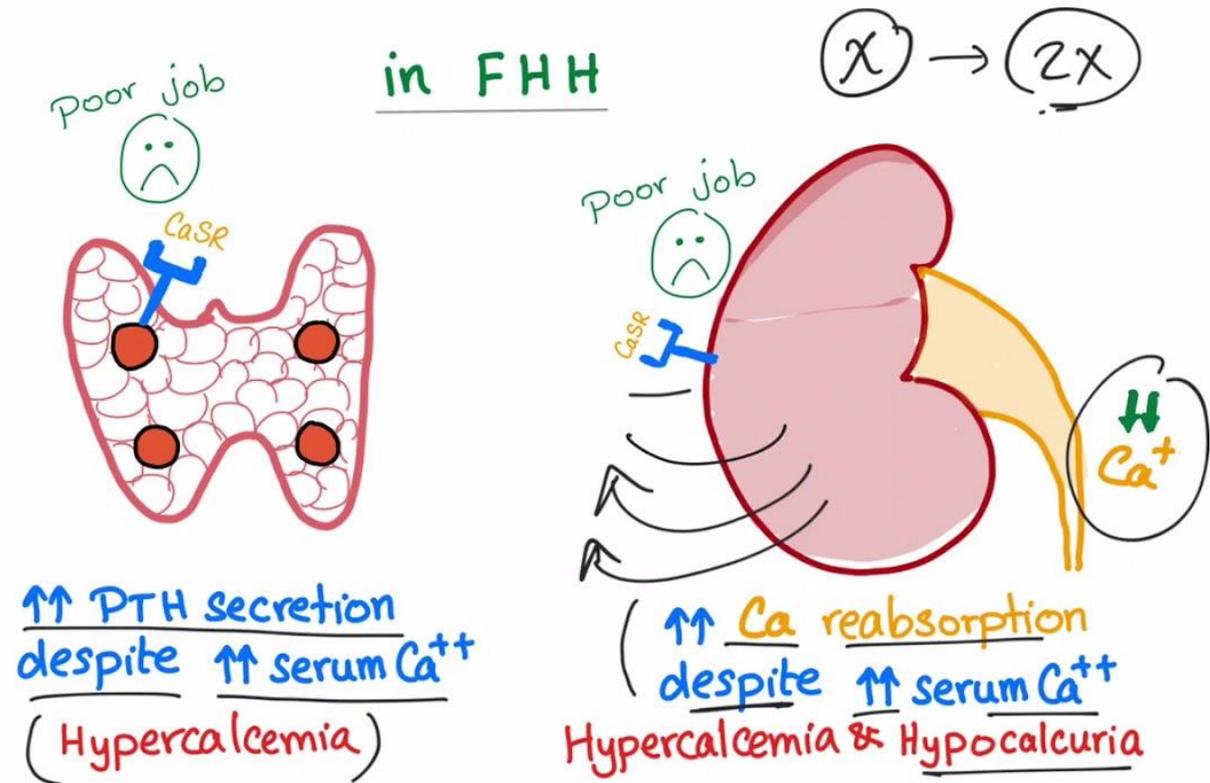
Hyperparathyroidism

Primary, Tertiary

Ectopic

Hypocalcemic hypercalcemia

? CaSR

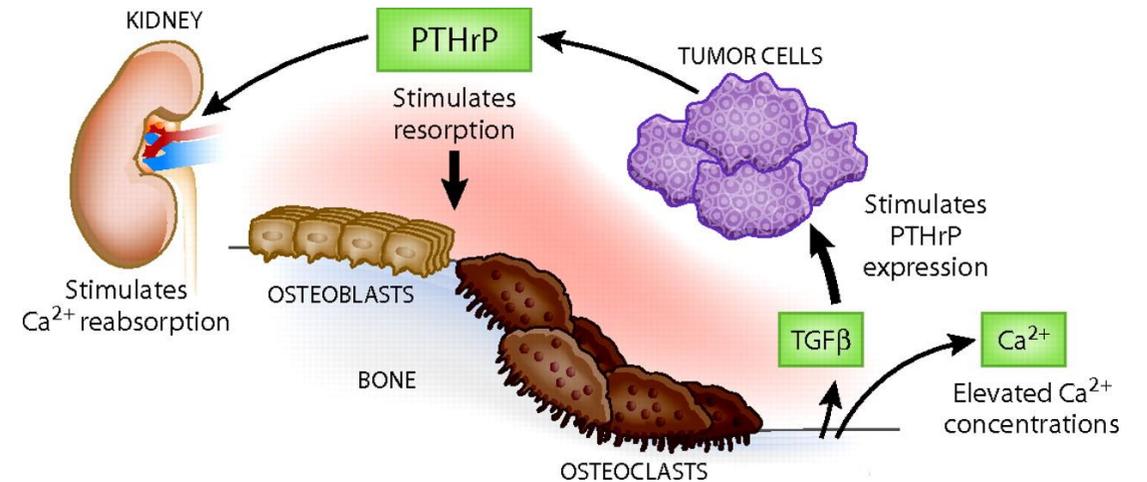


Hypercalcemia of malignancy

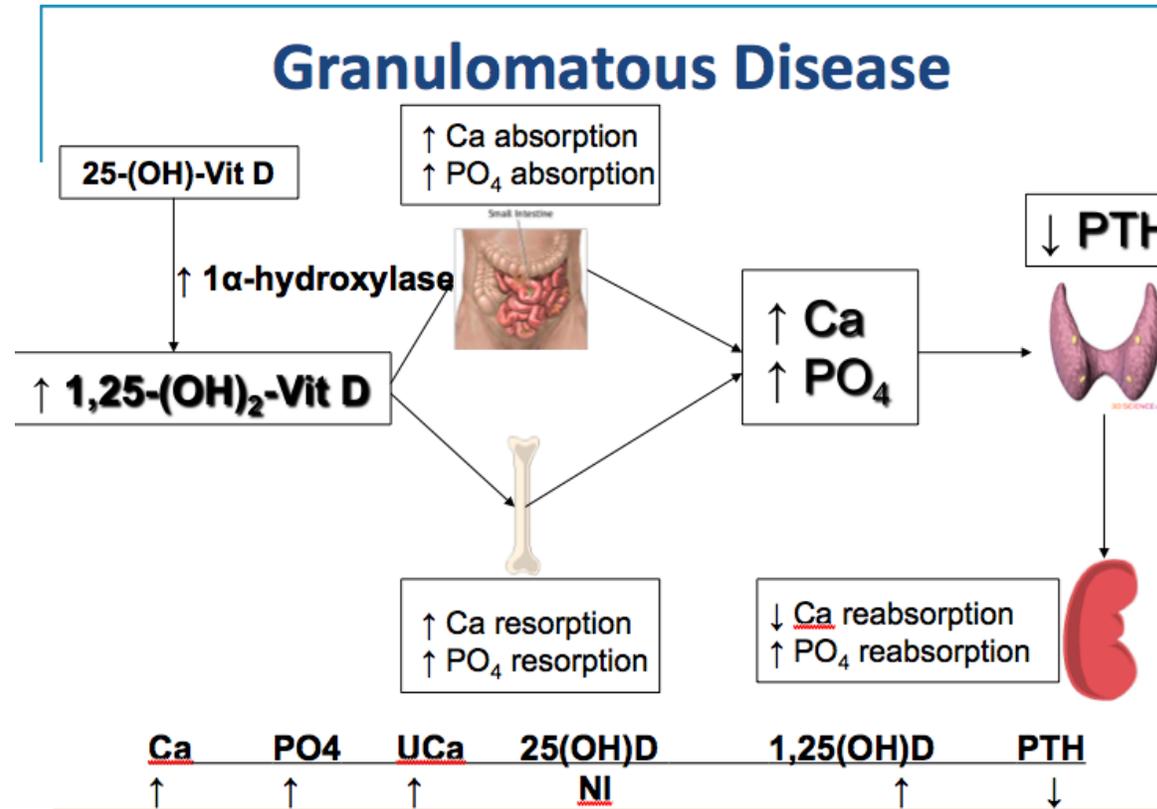
Types of Hypercalcemia of Malignancy

Table 1. Types of Hypercalcemia Associated with Cancer.*

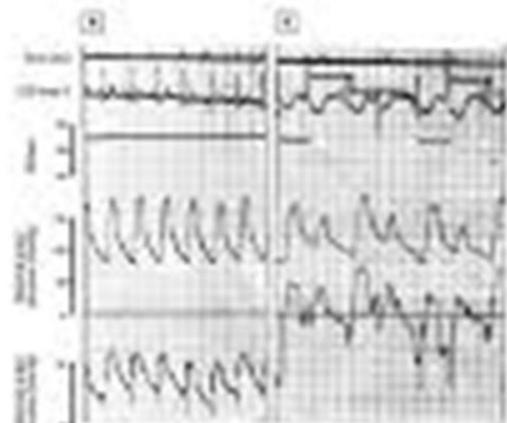
Type	Frequency (%)	Bone Metastases	Causal Agent	Typical Tumors
Humoral hypercalcemia of malignancy	80	Minimal or absent	PTHrP	Squamous cell cancer Renal cancer, Ovarian cancer, Endometrial cancer, ATLL, Breast cancer
Local osteolytic hypercalcemia	20	Common, extensive	Cytokines, chemokines, PTHrP	Breast cancer, multiple myeloma, lymphoma



Hypercalcemia of granulomatous disorders



CLINICAL MANIFESTATIONS OF HYPERCALCEMIA



Cardiac Dysrhythmias



Mental status changes:
lethargy, confusion,
memory loss



Decreased GI
Motility



Constipation



Nausea



Vomiting

Diagnostic approach to Hypercalcemia

Albumin concentrations

A detailed history

Serum PTH level and serum Phosphate

Serum creatinine and calcium/creatinine clearance ratio(<0.01 is suggestive of FHH)

sequence analysis of the CaSR gene

? A suppressed PTH level in the face of hypercalcemia

PTHrP

Serum 1,25(OH)₂D levels are increased in granulomatous disorders

Treatment

Volume expansion

Bisphosphonates

Hydrocortisone

Hypocalcemia

TABLE 65-2 CAUSES OF HYPOCALCEMIA

Low Parathyroid Hormone Levels (Hypoparathyroidism)

Parathyroid agenesis

- Isolated
- DiGeorge's syndrome

Parathyroid destruction

- Surgical
- Radiation
- Infiltration by metastases or systemic diseases
- Autoimmune

Reduced parathyroid function

- Hypomagnesemia
- Activating CaSR or G protein mutations

High Parathyroid Hormone Levels (Secondary Hyperparathyroidism)

Vitamin D deficiency or impaired 1,25(OH)₂D production/action

- Nutritional vitamin D deficiency (poor intake or absorption)
- Renal insufficiency with impaired 1,25(OH)₂D production
- Vitamin D resistance, including receptor defects

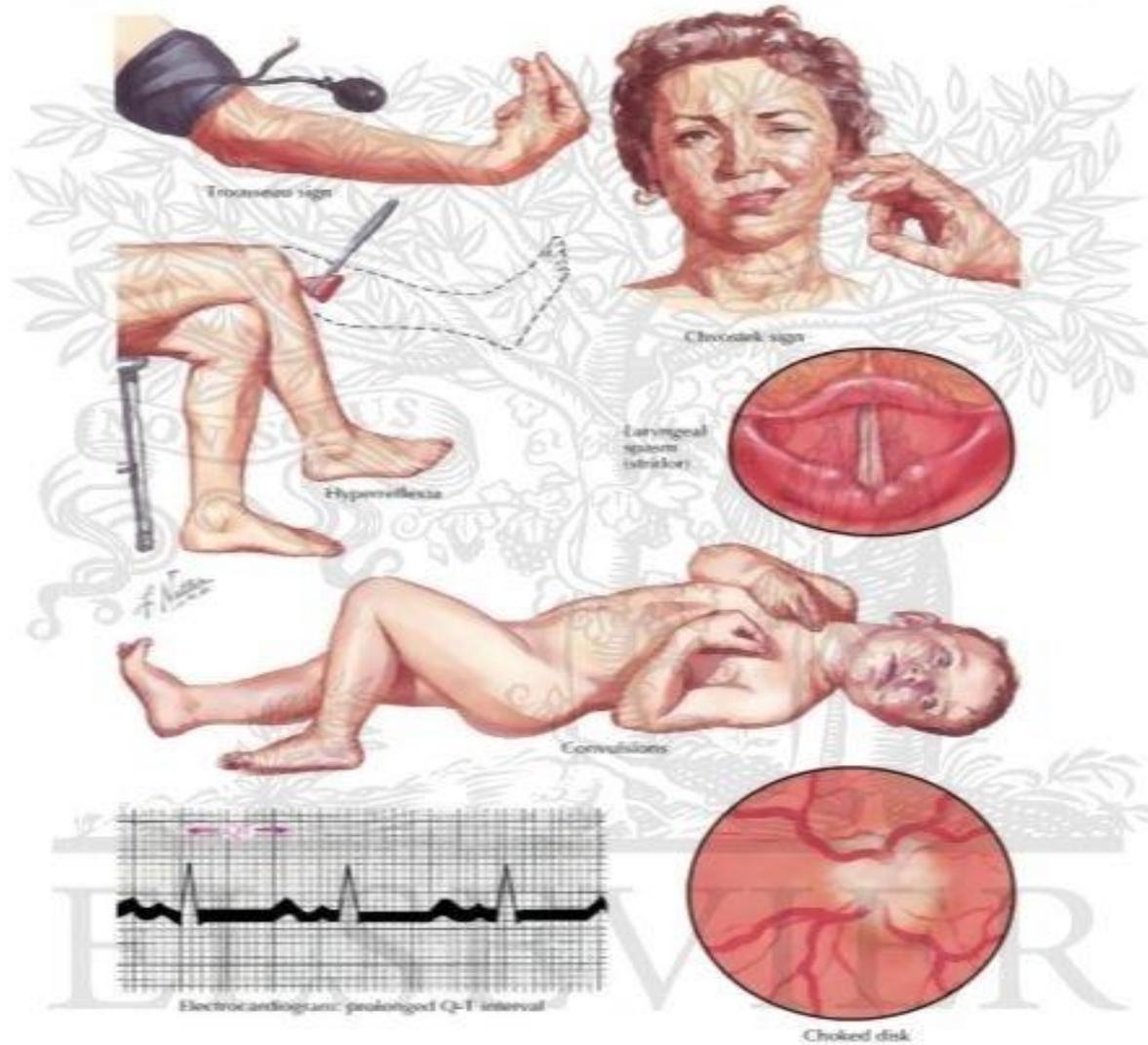
Parathyroid hormone resistance syndromes

- PTH receptor mutations
- Pseudohypoparathyroidism (G protein mutations)

Drugs

- Calcium chelators
- Inhibitors of bone resorption (bisphosphonates, plicamycin)
- Altered vitamin D metabolism (phenytoin, ketoconazole)

Tetany



Diagnostic evaluation

Albumin, Phosphorus and Magnesium levels

PTH level

History

serum 25-hydroxyvitaminD levels

Treatment

Iv Ca gluconate

Oral Calcium and vitamin D supplements

Treatment

Acute

i.v. calcium gluconate

Magnesium supplementation

Chronic

Calcium supplements and either vitamin D, or D3

*Thank
you*

